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A.P. Chemistry Practice Test - Ch. 13: Equilibrium ...

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Chapter 13 - Chemical Equilibrium | CourseNotes

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Chapter 13 - (Properties of Solutions)

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Ch. 13: Equilibrium

Name ____ MULTIPLE

CHOICE. Choose the one

alternative that best

completes the statement

or answers the question.

1) At equilibrium, ____.

A) the rates of the forward

and reverse reactions are

equal B) the rate constants

of the forward and

reverse reactions are

equal A.P. Chemistry

Practice Test - Ch. 13:

Equilibrium ...8) If this system is at equilibrium in a closed vessel & a small amount of H₂O is added, what will happen to the temperature inside the vessel? $\text{N}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{l}) \leftrightarrow \text{N}_2\text{H}_4(\text{g}) + \text{HNO}_2(\text{g})$ $\Delta H = -545 \text{ kJ/mol rxn}$ 9) $K_c = 3.2$ for this reaction: $\text{C}(\text{s}) + \text{CO}_2 \leftrightarrow 2 \text{CO}(\text{g})$. The concentration of CO in equilibrium with 0.50 M CO₂ is _____. 10) If the reaction flask is placed into an ice bath ...AP Chemistry Test (Chapter 13) - Denton ISD Start studying AP Chemistry: Chapter 13 Solutions. Learn vocabulary, terms, and

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Chemistry Test (Chapter 13) Multiple Choice (20%) 1 ...A.P. Chemistry Practice Test: Ch. 12, Kinetics MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1) Consider the following reaction: $3\text{A} \rightarrow 2\text{B}$ The average rate of appearance of B is given by $D[\text{B}]/Dt$. Comparing the rate of appearance of B and the rate ofA.P. Chemistry Practice Test: Ch. 12, Kinetics MULTIPLE ...These are the answers and explanations to the practice test on Chapters 1 - 3, which can be found here: <https://goo.gl/NgVq75> Chapters 1 - 3 Practice Test- Section summaries - Practice exercises - Practice test questions (Although we skipped a lot of the content in the text for Chapter 12, most of the practice test questions are in line with the AP Chemistry curriculum.)Course: AP Chemistry - Mr. von WerderAP Chemistry is an in-depth, fast-paced second-year chemistry course for advanced, science-oriented students. The course will provide students with a thorough grounding in chemical principles and quantitative reasoning,

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Chapter 13 - (Properties of Solutions)

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AP Chemistry: Chapter 13-

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